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Stoichiometry Gizmo Quiz Answers

The Stoichiometry Gizmo™ allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction. To begin, check that this equation is shown: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$. Look at the coefficients (such as the “3” in 3CO) in front of each substance in the equation.

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Solve problems in chemistry using dimensional analysis. Select appropriate tiles so that units in the question are converted into units of the answer. Tiles can be flipped, and answers can be calculated once the appropriate unit conversions have been applied.

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Since 20 moles NO₂ are produced, and it takes 3 moles O₂ to produce 4 moles of NO₂, the answer would be 15 moles. $15 \text{ moles} = 20 \text{ moles NO}_2 \times (3 \text{ moles O}_2 / 4 \text{ moles NO}_2)$ --- This is the molar ratio between the two.

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