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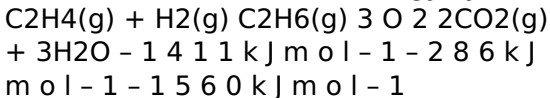
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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle.



(hydrogenation) (l) Calculate the value

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for the enthalpy of hydrogenation of ethene from the energy cycle.....

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8 20 2 2 2 2 0 correct reactants and products; [2] correct coefficients; M2 can only be scored if M1 correct. (b) Local pollutant: carbon monoxide/CO / volatile organics/VOCs / nitrogen oxide/NO / (unburnt) hydrocarbons; Do not accept

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methane/ CH_4 , ethane/ C_2H_6 ,
propane/ C_3H_8 or butane/ C_4H_{10} .
Global pollutant: nitrogen oxide/ NO /
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$\text{C}_2\text{H}_5\text{OH}(\text{l}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 3\text{H}_2\text{O}(\text{l})$ $\Delta H = -1367 \text{ kJ mol}^{-1}$
 $\text{C}_8\text{H}_{18}(\text{l}) + 12.5\text{O}_2(\text{g}) \rightarrow 8\text{CO}_2(\text{g}) + 9\text{H}_2\text{O}(\text{l})$ $\Delta H = -5470 \text{ kJ mol}^{-1}$

(i) Calculate the mass, in g, of ethanol and octane in 1.00 kg of the fuel mixture. [1].....

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cause pain/prostaglandins; strong analgesics work by bonding to receptor sites in the brain / prevent the transmission of pain impulses without depressing the central nervous system;
[2]

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